

### **Our Mission Continues**

We are proud to present once again our annual water quality report covering all testing performed between January 1 and December 31, 2014. Most notably, last year marked the 40th anniversary of the Safe Drinking Water Act (SDWA). This rule was created to protect public health by regulating the nation's drinking water supply. We celebrate this milestone as we continue to manage our water system with a mission to deliver the best quality drinking water. By striving to meet the requirements of SDWA, we are ensuring a future of healthy, clean drinking water for years to come.

Please let us know if you ever have any questions or concerns about your water.

# Important Health Information

Some people may be more vulnerable to contaminants in drinking water than the general population. Immunocompromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants may be particularly at risk from infections. These people should seek advice about drinking water from their health care providers. The U.S. EPA/CDC (Centers for Disease Control and Prevention) guidelines on appropriate means to lessen the risk of infection by *Cryptosporidium* and other microbial contaminants are available from the Safe Drinking Water Hotline at (800) 426-4791 or http://water.epa.gov/drink/hotline.

# Community Participation

You are invited to participate in our public forum and voice your concerns about your drinking water. We meet the first and third Mondays of each month beginning at 6:00 p.m. at the Claude L Wells Education Center at 300 S. Robinson St.



## Substances That Could Be in Water

The sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land or through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

In order to ensure that tap water is safe to drink, the U.S. Environmental Protection Agency (U.S. EPA) and the State Water Resources Control Board (State Board) prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. State Board regulations also establish limits for contaminants in bottled water that must provide the same protection for public health. Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk.

Contaminants that may be present in source water include:

**Microbial Contaminants**, such as viruses and bacteria, that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife;

**Inorganic Contaminants**, such as salts and metals, that can be naturally occurring or can result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming;

**Pesticides and Herbicides**, that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses;

**Organic Chemical Contaminants**, including synthetic and volatile organic chemicals, which are by-products of industrial processes and petroleum production, and which can also come from gas stations, urban stormwater runoff, agricultural applications, and septic systems;

**Radioactive Contaminants**, that can be naturally occurring or can be the result of oil and gas production and mining activities.

More information about contaminants and potential health effects can be obtained by calling the U.S. EPA's Safe Drinking Water Hotline at (800) 426-4791.

# Lead in Home Plumbing

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. We are responsible for providing high-quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at www.epa.gov/safewater/lead.

# Where Does My Water Come From?

The City of Tehachapi uses only groundwater pumped from the Tehachapi Basin aquifer; no surface or imported water is used for direct consumption. Six active deep wells within the city continually refill 5 million gallons of storage facilities and the 40 miles of transmission lines that bring water to the homes, schools, and businesses served by our system.

The City operates five pressure zones, four of which are used and tested. Monthly bacteriological testing is done in all four zones as well as in the storage tanks and wells themselves. A free chlorine residual of 0.21 - 2.14 mg/l (parts per million) is maintained throughout the distribution system.

### Water System Information

Of the six active wells operated by the city, one is equipped with standby power for use in case of an emergency. These wells are valved so that water can be diverted in different directions in the event of a catastrophic line rupture. The City also has a portable generator for use at a second well or at the booster station located at the Curry Street Tank Site.

### **Testing**

The City of Tehachapi performs water quality testing in accordance with all federal and state criteria. Although comprehensive testing was done in 2012, only detected contaminants will be reported in this report.

The City's water sampling (both chemical and bacteriological) is done by a state-certified water treatment plant operator and analyzed by a state-certified laboratory to ensure accuracy in testing.

### Water Conservation

You can play a role in conserving water and saving yourself money in the process by becoming conscious of the amount of water your household is using and by looking for ways to use less whenever you can. It is not hard to conserve water. Here are a few tips:

- Automatic dishwashers use 15 gallons for every cycle, regardless of how many dishes are loaded.
  So get a run for your money and load it to capacity.
- Turn off the tap when brushing your teeth.
- Check every faucet in your home for leaks. Just a slow drip can waste 15 to 20 gallons a day. Fix it and you can save almost 6,000 gallons per year.
- Check your toilets for leaks by putting a few drops of food coloring in the tank. Watch for a few minutes to see if the color shows up in the bowl. It is not uncommon to lose up to 100 gallons a day from an invisible toilet leak. Fix it and you save more than 30,000 gallons a year.
- Use your water meter to detect hidden leaks. Simply turn off all taps and water using appliances. Then check the meter after 15 minutes. If it moved, you have a leak.

# QUESTIONS?

For more information about this report, or for any questions relating to your drinking water, please call Jon Curry, Public Works Director, at (661) 822-4078 ext. 201, or Thomas Brown, WTPO, at (661) 822-4078 ext. 203.

## To The Last Drop

The National Oceanic and Atmospheric Administration (NOAA) defines drought as a deficiency in precipitation over an extended period of time, usually a season or more, resulting in a water shortage causing adverse impacts on vegetation, animals, and/or people. Drought strikes in virtually all climate zones, from very wet to very dry.

There are primarily three types of drought: Meteorological Drought refers to the lack of precipitation, or the degree of dryness and the duration of the dry period; Agricultural



Drought refers to the agricultural impact of drought, focusing on precipitation shortages, soil water deficits, and reduced ground water or reservoir levels needed for irrigation; and Hydrological Drought, which pertains to drought that usually occurs following periods of extended precipitation shortfalls that can impact water supply (i.e., stream flow, reservoir and lake levels, ground water).

Drought is a temporary aberration from normal climatic conditions, thus it can vary significantly from one region to another. Although normally occurring, human factors, such as water demand, can exacerbate the duration and impact that drought has on a region. By following simple water conservation measures, you can help significantly reduce the lasting effects of extended drought.

To learn more about water conservation efforts, check out U.S. EPA's Water Conservation Tips for Residents at www.epa.gov/region1/eco/drinkwater/water\_conservation\_residents.html.

# What's Your Water Footprint?

You may have some understanding about your carbon footprint, but how much do you know about your water footprint? The water footprint of an individual, community, or business is defined as the total volume of freshwater that is used to produce the goods and services that are consumed by the individual or community or produced by the business. For example, 11 gallons of water are needed to irrigate and wash the fruit in one half-gallon container of orange juice. Thirty-seven gallons of water are used to grow, produce, package, and ship the beans in that morning cup of coffee. Two hundred and sixty-four gallons of water are required to produce one quart of milk, and 4,200 gallons of water are required to produce two pounds of beef.

According to the U.S. EPA, the average American uses over 180 gallons of water daily. In fact, in the developed world, one flush of a toilet uses as much water as the average person in the developing



world allocates for an entire day's cooking, washing, cleaning, and drinking. The annual American per capita water footprint is about 8,000 cubic feet; twice the global per capita average. With water use increasing six-fold in the past century, our demands for freshwater are rapidly outstripping what the planet can replenish.

To check out your own water footprint, go to www.gracelinks.org/824/water-program or visit www. waterfootprint.org to see how the water footprints of other nations compare.

### Source Water Assessment

The City of Tehachapi conducted a water source assessment and protection program. The assessment for the Mojave Well identified vulnerabilities from activities located near the drinking water source. The source is considered most vulnerable to sewer collection systems and to a historic gas station within the five-and ten-year times of travel. The source has a 100-foot sanitary seal and a depth of 182 feet to the uppermost perforation. Any microbiological activity would have to travel this vertical distance to the aquifer before it could begin horizontal travel to the well. The gas station has not had any problems associated with it, and no gas products have ever been detected in Mojave Well.

For Dennison Well, again, no contaminants above the MCL have been detected in the water supply; however, the assessment identified vulnerabilities from activities located nearby. These vulnerabilities include high-density housing and the close proximity of other supply wells, which violate specifications requiring distances far enough so that contaminants would take a minimum of two years to reach the water supply. Both of these vulnerabilities pose a relatively low-ranking risk, as does potential leaching from gas stations—both active and historic—and confirmed leaking from a tank within the 10-year time of travel. Snyder Well is considered most vulnerable to sewer collection systems. Snyder Well is a standby source and was not used for water production in 2014.

No contaminants above the MCL have been detected in the water supplied from Curry Well. The assessment noted that the water supply is still considered vulnerable to activities located near the drinking water source.

Minton Well's supply was assessed and no contaminants above the MCL were found, although it is still considered vulnerable to activities located near the drinking water source.

No contaminants above the MCL have been detected in the water supplied from Wahlstrom Well. The assessment considers the source to be vulnerable to activities located near the drinking water supply.

Pinon Well is considered most vulnerable to septic systems—both low-density and sewer collection systems. No contaminants above the MCL have been detected in the water supply; however, the source is considered vulnerable to activities located near the drinking water source. This source has a very deep 300-foot sanitary seal. In addition, the depth to the uppermost perforation is 400 feet. Any microbiological activity would have to travel this vertical distance to the aquifer before it could begin horizontal travel to the well. A copy of the complete assessment may be viewed at the City of Tehachapi, 115 South Robinson Street, Tehachapi, CA 93561.

## Sampling Results

During the past year we have taken hundreds of water samples in order to determine the presence of any radioactive, biological, inorganic, volatile organic or synthetic organic contaminants. The table below shows only those contaminants that were detected in the water. The state requires us to monitor for certain substances less than once per year because the concentrations of these substances do not change frequently. In these cases, the most recent sample data are included, along with the year in which the sample was taken.

| REGULATED SUBSTANCES                            |                 |                   |                          |                    |                   |           |  |
|---|-----------------|-------------------|--------------------------|--------------------|-------------------|-----------|--|
| SUBSTANCE<br>(UNIT OF MEASURE)                  | YEAR<br>SAMPLED | MCL<br>[MRDL]     | PHG<br>(MCLG)<br>[MRDLG] | AMOUNT<br>DETECTED | RANGE<br>LOW-HIGH | VIOLATION | TYPICAL SOURCE   |
| Chlorine (ppm)                                  | 2014            | [4.0 (as<br>Cl2)] | [4 (as<br>Cl2)]          | 1.09               | 0.28-1.88         | No        | Drinking water disinfectant added for treatment  |
| Fluoride (ppm)                                  | 2012            | 2.0               | 1                        | 0.27               | 0.12-0.55         | No        | Erosion of natural deposits; water additive<br>that promotes strong teeth; discharge from<br>fertilizer and aluminum factories |
| Gross Alpha Particle Activity (pCi/L)           | 2006            | 15                | (0)                      | 1.31               | 0.22-2.5          | No        | Erosion of natural deposits  |
| Nitrate [as nitrate] <sup>1</sup> (ppm)         | 2014            | 45                | 45                       | 29                 | 6.2–44            | No        | Runoff and leaching from fertilizer use;<br>leaching from septic tanks and sewage; erosion<br>of natural deposits              |
| TTHMs [Total Trihalomethanes]-<br>Stage 2 (ppb) | 2014            | 80                | NA                       | 1.06               | 0.52–1.5          | No        | By-product of drinking water disinfection  |

Tap water samples were collected for lead and copper analyses from sample sites throughout the community

| SUBSTANCE<br>(UNIT OF MEASURE) | YEAR<br>SAMPLED | AL  | PHG<br>(MCLG) | AMOUNT<br>DETECTED<br>(90TH%TILE) | SITES ABOVE<br>AL/TOTAL<br>SITES | VIOLATION | TYPICAL SOURCE  |
|--------------------------------|-----------------|-----|---------------|-----------------------------------|----------------------------------|-----------|---|
| Copper (ppm)                   | 2013            | 1.3 | 0.3           | 0.27                              | 0/20                             | No        | Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives                     |
| Lead (ppb)                     | 2013            | 15  | 0.2           | 4.4                               | 0/20                             | No        | Internal corrosion of household water plumbing systems;<br>discharges from industrial manufacturers; erosion of natural<br>deposits |

### **SECONDARY SUBSTANCES**

| SUBSTANCE<br>(UNIT OF MEASURE) | YEAR<br>SAMPLED | SMCL  | PHG<br>(MCLG) | AMOUNT<br>DETECTED | RANGE<br>LOW-HIGH | VIOLATION | TYPICAL SOURCE  |
|--------------------------------|-----------------|-------|---------------|--------------------|-------------------|-----------|---|
| Manganese (ppb)                | 2012            | 50    | NS            | 10                 | 10–10             | No        | Leaching from natural deposits                              |
| Specific Conductance (µS/cm)   | 2012            | 1,600 | NS            | 484                | 413–534           | No        | Substances that form ions when in water; seawater influence |
| Sulfate (ppm)                  | 2012            | 500   | NS            | 39.8               | 25–63             | No        | Runoff/leaching from natural deposits; industrial wastes    |
| Total Dissolved Solids (ppm)   | 2012            | 1,000 | NS            | 303                | 260-340           | No        | Runoff/leaching from natural deposits                       |

#### **OTHER SUBSTANCES**

| SUBSTANCE<br>(UNIT OF MEASURE) | YEAR<br>SAMPLED | AMOUNT<br>DETECTED | RANGE<br>LOW-HIGH | TYPICAL SOURCE                  |
|--------------------------------|-----------------|--------------------|-------------------|---------------------------------|
| Bicarbonate (ppm)              | 2012            | 183                | 160–220           | Leaching from natural deposits  |
| Calcium (ppm)                  | 2012            | 58.5               | 40–72             | Generally found in ground water |
| pH (Units)                     | 2012            | 7.92               | 7.85–8.06         | Naturally occurring             |
| Potassium (ppm)                | 2012            | 1.2                | 1.0-1.6           | Generally found in ground water |
| Total Hardness (ppm)           | 2012            | 176                | 120-220           | Naturally occurring             |

Nitrate in drinking water at levels above 45 ppm is a health risk for infants of less than six months of age. Such nitrate levels in drinking water can interfere with the capacity of the infant's blood to carry oxygen, resulting in a serious illness; symptoms include shortness of breath and blueness of the skin. Nitrate levels above 45 ppm may also affect the ability of the blood to carry oxygen in other individuals, such as pregnant women and those with certain specific enzyme deficiencies. If you are caring for an infant, or you are pregnant, you should ask advice from your health care provider.

### **Definitions**

AL (Regulatory Action Level): The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**μS/cm (microsiemens per centimeter):** A unit expressing the amount of electrical conductivity of a solution.

MCL (Maximum Contaminant Level): The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs (SMCLs) are set to protect the odor, taste and appearance of drinking water.

MCLG (Maximum Contaminant Level Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. EPA.

MRDL (Maximum Residual Disinfectant Level): The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

MRDLG (Maximum Residual Disinfectant Level Goal): The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

NA: Not applicable

NS: No standard

**pCi/L** (**picocuries per liter**): A measure of radioactivity.

**PDWS** (Primary Drinking Water Standard): MCLs and MRDLs for contaminants that affect health along with their monitoring and reporting requirements, and water treatment requirements.

**PHG** (Public Health Goal): The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California EPA.

**ppb** (parts per billion): One part substance per billion parts water (or micrograms per liter).

**ppm (parts per million):** One part substance per million parts water (or milligrams per liter).